[Total No. of Questions - 9] [Total No. of Printed Pages - 3] (2064)

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B. Tech 2nd Semester Examination Applied Chemistry (O.S.) AS-1004

Time: 3 Hours Max. Marks: 100

The candidates shall limit their answers precisely within the answerbook (40 pages) issued to them and no supplementary/continuation sheet will be issued.

Note: Attempt five question in all selecting one question from each section A, B, C and D. Section E is compulsory.

SECTION - A

(a) Give the statements of second law of thermodynamics.

 (4)

 (b) Derive Gibb's Helmholtz equation.

 (6)

 (c) Derive the conditions for spontaniety of a process in terms of entropy.

 (6)

 (d) Gibb's free energy of a reaction at 27°C and 37°C are -29.0 Kcal and -10 Kcal respectively. Calculate the free energy at 330K.

 (4)

- 2. (a) Explain the following terms: (i) Phase, (ii) Component, (iii) Degree of freedom. (6)
 - (b) Discuss the phase diagram of water system. (8)
 - (c) Write short notes on: (i) Eutectic paint, (ii) Reduced phase rule. (6)

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SECTION - B

3.	(a) Discuss the ion-exchange process for water soft		ng.
			(6)
	(b)	Distinguish between hard water and soft water. W	hat is
		break point chlorination?	(6)

- (c) Write short note on Cottrele's electrostatic precipitator. (4)
- (d) Write short note on: (i) Ozone depletion, (ii) greenhouse effect. (4)
- (a) Define corrosion of metals? What are different types of corrosion? Explain electrochemical theory of wet corrosion.
 (8)
 - (b) Write short notes on: (i) Pitting corrosion, (ii) Role of sacraficial anode in corrosion controls. (6)
 - (c) Give an account of tinning and galvanising. (6)

SECTION - C

- 5. (a) Define the term lubrication and lubricants. What are different types of lubricants. (10)
 - (b) What are greases and under what situation are they employed? Discuss the composition and uses of:
 - (i) Calcium based, (ii) Soda-based, (iii) axle grease. (10)
- 6. (a) What is meant by cracking of petroleum? Explain fluidized-bed catalytic method of obtaining gasoline. Give its mechanisms. (10)
 - (b) How can power alcohol be useful in fuel crisis? (10)

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SECTION - D

- 7. (a) What are Miller indices? Calculate the Miller indices of a crystal plane which cut the cell axis $\left(\frac{1}{2}a,\,\frac{2}{3}b,\,\infty\right)$. (10)
 - (b) What are instrinsic semiconductor? Explain the conduction in n-type and p-type semiconductor. (10)
- 8. (a) Explain the following:
 - (i) Auto catalysis, (ii) Promoters, (iii) Negative catalysis, (iv) Acid-base catalyst, (v) Enzyme catalysis. (10)
 - (b) Write short note on homogeneous and heterogeneous catalysis. (10)

SECTION - E

- 9. (a) What is ppm?
 - (b) Distinguish between gross and net calorific value of fuel.
 - (c) What is meant by knocking?
 - (d) What is BOD and COD?
 - (e) Name the three gases in atmosphere which cause acid rain.
 - (f) Give an application of Pb-Ag system.
 - (g) What type of lubricants are used for transformers?
 - (h) Which of the following three conditions is valid for spontaneous reaction?
 - (i) $\Delta G > 0$, (ii) $\Delta G = 0$, (iii) $\Delta G < 0$
 - (i) Why is germanium doped with antimony called an n-type semiconductor?
 - (j) What is a super conductor? (2×10=20)