[Total No. of Questions - 7] [Total No. of Printed Pages - 2] (2064)

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B. Pharmacy (Ayurveda) 2nd Semester Examination Pharmaceutical Chemistry (Inorganic) (N.S.) BPA-222

Time: 3 Hours Max. Marks: 70

The candidates shall limit their answers precisely within the answerbook (40 pages) issued to them and no supplementary/continuation sheet will be issued.

Note: Question 1 is compulsory and candidates are required to attempt any FIVE questions out of remaining six.

- 1. Explain the following:
 - (a) Equivalent point
 - (b) Electromagnetic spectrum
 - (c) Pauli's exclusion principle
 - (d) Quantum number
 - (e) Aufbau principle
 - (f) Non-aqueous titration
 - (g) Principle for Gravimetric analysis
 - (h) Ritz combination principle
 - (i) Photoelectric effect
 - (j) Radial probability distribution curve for 1S electron. (20)
- 2. Enumerate types of titrametric methods of analysis and discuss on replacement titration method with example. (10)

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- 3. Write on preparation and uses of Ammonium chloride. (10)
- 4. Explain the defects in Mendeleev's periodic table and explain how the modern periodic table remove them. (10)
- 5. Discuss occurrence, properties and reactions of sodium. Write preparation, properties and uses of sodium chloride. (10)
- 6. Write on quantification of heavy metals in Ayurvedic preparations. (10)
- 7. Discuss reactions of metals with acid, oxygen and water with examples. (10)