[Total No. of Questions - 18] [Total No. of Printed Pages - 2] (2125)

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B. Pharmacy 3rd Semester Examination Physical Pharmacy-I (NS) BP-233

Time: 3 Hours Max. Marks: 70

The candidates shall limit their answers precisely within the answerbook (40 pages) issued to them and no supplementary/continuation sheet will be issued.

SECTION - A (Attempt any two questions)

- 1. What are accelerated stability studies? What are the salient features & limitations of the accelerated stability studies?
- 2. Define order of reaction. Explain the different methods for determination of order of reaction.
- 3. What is critical constant? Explain the different methods of liquefaction of gases. (2×10=20)

SECTION - B (Attempt any eight questions)

- 4. What is partition coefficient? Give its application in pharmacy.
- 5. Explain in detail about Triple Point.
- 6. Describe the influence of temperature on the rate of reaction.
- 7. Write comparison between Isothermal and Adiabatic expansion.
- 8. Elaborate on colligative properties and Eutectic mixtures.
- 9. Write a note on pharmaceutical buffers.
- 10. Discuss the applications of the phase rule.

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- 11. Briefly discuss the solubility of gases in liquids.
- 12. Write a note on chemical degradation of drugs.
- 13. Write short notes on:
 - (a) Free energy functions and its measurement.
 - (b) Half-life determination. (8×5=40)

SECTION - C (All questions are compulsory)

- 14. Explain Debye Huckle theory.
- 15. What is meant by physical degradation of drugs?
- 16. Define ideal and real solutions.
- 17. How crystalline solid differs from amorphous solids?
- 18. Write a note on Refractive Index and Molar Refraction.

 $(5 \times 2 = 10)$